

FORMULA

EXERCISE

II

1. potassium chloride <i>KCl</i>	31. sodium carbonate <i>Na₂CO₃</i>	61. sodium bicarbonate <i>Na</i>
2. lead sulphide <i>PbS</i>	32. zinc acetate <i>Zn(C₂H₃O₂)₂</i>	62. sulphuric acid <i>H₂SO₄</i>
3. zinc oxide <i>ZnO</i>	33. lead acetate <i>Pb(C₂H₃O₂)₂</i>	63. potassium nitrite <i>KNO₂</i>
4. aluminum chloride <i>AlCl₃</i>	34. magnesium nitrate <i>Mg(NO₃)₂</i>	64. sodium perchlorate <i>NaClO₄</i>
5. magnesium bromide <i>MgBr</i>	35. ammonium acetate <i>CH₃COONH₄</i>	65. nitrous acid <i>HNO₂</i>
6. phosphorus iodide <i>PI</i>	36. potassium bromate <i>KBrO₃</i>	66. zinc hypochlorite <i>Zn(ClO)₂</i>
7. calcium nitride <i>Ca₃N₂</i>	37. sodium chlorate <i>NaClO₃</i>	67. lead periodate <i>Pb(IO₄)₂</i>
8. silver iodide <i>AgI</i>	38. ammonium carbonate <i>(NH₄)₂CO₃</i>	68. silver bromite <i>AgBrO</i>
9. silver sulphide <i>Ag₂S</i>	39. aluminum phosphate <i>AlPO₄</i>	69. sodium potassium ammonium phosphate <i>NaK₂AlPO₄</i>
10. aluminum oxide <i>Al₂O₃</i>	40. zinc sulphate <i>ZnSO₄</i>	70. ammonium potassium sodium phosphate <i>Na₂AlK₂PO₄</i>
11. potassium sulphide <i>K₂S</i>	41. potassium carbonate <i>K₂CO₃</i>	71. sodium peroxide <i>Na₂O₂</i>
12. calcium chloride <i>CaCl₂</i>	42. sodium cyanide <i>NaCN</i>	72. sulphurous acid <i>H₂SO₃</i>
13. sodium iodide <i>NaI</i>	43. aluminum sulphate <i>Al₂(SO₄)₃</i>	73. hydrochloric acid <i>HCl</i>
14. carbon dioxide <i>CO₂</i>	44. ammonium phosphate <i>(NH₄)₂PO₄</i>	74. perchloric acid <i>HClO₄</i>
15. sulphur trioxide <i>SO₃</i>	45. calcium carbonate <i>CaCO₃</i>	75. sodium hypobromite <i>NaBrO</i>
16. lead bromide <i>PbBr₂</i>	46. magnesium acetate <i>Mg(CH₃COO)₂</i>	76. hydrocyanic acid <i>HCN</i>
17. potassium phosphide <i>K₃P</i>	47. silver nitrate <i>AgNO₃</i>	77. barium peroxide <i>BaO₂</i>
18. aluminum nitride <i>AlN</i>	48. calcium phosphate <i>Ca₃(PO₄)₂</i>	78. calcium sulphite <i>CaSO₃</i>
19. zinc sulphide <i>ZnS</i>	49. ammonium sulphate <i>(NH₄)₂SO₄</i>	79. acetic acid <i>CH₃COOH</i>
20. magnesium phosphide <i>Mg₃P₂</i>	50. zinc iodate <i>ZnIO₃</i>	80. ammonia <i>NH₃</i>
21. ferrous chloride <i>FeCl₂</i>	51. ammonium cyanide <i>NH₄CN</i>	81. calcium iodite <i>CaI₂</i>
22. mercuric oxide <i>HgO</i>	52. ammonium nitride <i>NH₄N₃</i>	82. zinc phosphite <i>Zn₃(PO₄)₂</i>
23. stannous sulphide <i>SnS</i>	53. silver acetate <i>AgC₂H₃O₂</i>	83. ferrous hypochlorite <i>Fe(OH)Cl</i>
24. cupric bromide <i>CuBr₂</i>	54. ammonium nitrate <i>(NH₄)₂NO₃</i>	84. ammonium sulphite <i>NH₄HS</i>
25. stannic oxide <i>SnO₂</i>	55. aluminum bromate <i>Al(BrO₃)₃</i>	85. silver chloride <i>AgCl</i>
26. cuprous sulphide <i>Cu₂S</i>	56. potassium chlorate <i>KClO₃</i>	86. arsenic oxide <i>As₂O₃</i>
27. mercurous sulphide <i>HgS</i>	57. sodium nitrate <i>NaNO₃</i>	87. hydrogen peroxide <i>H₂O₂</i>
28. ferric oxide <i>Fe₂O₃</i>	58. lead sulphate <i>PbSO₄</i>	88. ammonium nitrite <i>(NH₄)₂NO₂</i>
29. ferrous oxide <i>Fe₂O₃</i>	59. ferric carbonate <i>Fe₂(CO₃)₃</i>	89. antimony sulphide <i>As₂S₃</i>
30. potassium iodide <i>KI</i>	60. stannic sulphate <i>SnSO₄</i>	90. cupric sulphate <i>CuSO₄</i>